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#### Pro-Solv 430 **GHS** Safety Data Sheet 24 Hour Emergency Telephone From: Performance Chemical CHEM-TEL, INC. 1-800-255-3924 9105 W Interstate 20 Midland, TX 79706 NOTE: CHEM-TEL emergency number to be Phone: 432-332-3059 used only in the event of chemical emergencies (432) 332-3097 involving a spill, leak, fire, exposure, or accident involving chemicals. All non-emergency questions should be directed to (432) 332-3059 for assistance.

# 1. Product Identification

**Trade Name** Pro-Solve 430 Paraffin Solvent / Dispersant

CAS Number: See Section 2
Product Family: Toluene
Synonyms: N/AP

#### 2. Hazards Identification

Hazards Classification: Flammable Liquid-Category 3

Skin Irritant-Category 2
Eye Irritant-Category 2a
TOST (Repeated)-Category 2
Aspiration Hazard-Category 2



#### **Hazard Statement:**

FLAMMABLE LIQUID – IRRITANT – ABSORBED THROUGH THE SKIN - CENTRAL NERVOUS SYSTEM -HARMFUL OR FATAL IF SWALLOWED - ASPIRATION HAZARD

#### **Precautionary Statement:**

Severe fire hazard. Vapor may cause flash fire or explosion! Keep away from heat, sparks, open flame, and other ignition sources.

If swallowed, DO NOT INDUCE VOMITING, as this may cause chemical pneumonia (fluid in the lungs). Contact with this product may cause eye, skin, and mucous membrane irritation. Harmful if absorbed through the skin. Avoid prolonged breathing of vapors or mists. Inhalation may cause irritation, anesthetic effects (dizziness, nausea, headache, intoxication) and respiratory system effects. Excessive exposure may affect the liver and kidneys. Prolonged and/or repeated inhalation may increase the heart's susceptibility to arrhythmias (irregular beats)!

# 3. Composition / Information on Ingredients

Component NameCAS Registry No.Concentration % (Wt.)Toluene108-88-320-40

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Heptane	142-82-5	20-50
Methanol	67-56-1	20-40

### 4. First Aid Measures:

Take proper precautions to ensure your own health and safety before attempting rescue or providing first aid. For more specific information, refer to Exposure Controls and Personal Protection in Section 8 of this MSDS.

Inhalation

Immediately move individual to fresh air. If individual is not breathing, immediately begin artificial respiration. If heart has stopped, immediately begin CardioPulmonary Resuscitation (CPR). If breathing is difficult, 100 percent humidified oxygen should be administered by a qualified individual. Seek medical attention immediately.

**Eye Contact** 

Check for and remove contact lenses. If irritation or redness develops, flush eyes with cool, clean low-pressure water for at least 15 minutes. Hold eyelids apart to ensure complete irrigation of the eye and eyelid tissue. Do not use eye ointment. Seek medical attention immediately.

**Skin Contact** 

Remove contaminated shoes and clothing. Flush affected area with large amounts of water. If skin surface is damaged, apply a clean dressing and seek medical attention. Do not use ointments. If skin surface is not damaged, clean affected area thoroughly with mild soap and water. Seek medical attention if tissue appears damaged or if pain or irritation persists.

Ingestion

DO NOT INDUCE VOMITING or give anything by mouth. If spontaneous vomiting is about to occur, place individual's head below his knees. If individual is drowsy or unconscious, place on his left side with head down. Never give anything by mouth to a person who is not fully conscious. Do not leave individual unattended. Seek medical attention immediately.

**Notes to Physician** 

Inhalation overexposure can produce toxic effects. Monitor for respiratory distress. If cough or difficulty in breathing develops, evaluate for upper respiratory tract inflammation, bronchitis, and pneumonitis. Vigorous anti-inflammatory / steroid treatment may be required at first evidence of upper airway or pulmonary edema. Administer 100 percent humidified supplemental oxygen with assisted ventilation as required.

If ingested, this material presents a significant aspiration / chemical pneumonitis hazard. As a result, induction of emesis is not recommended. Administer an aqueous slurry of activated charcoal followed by a cathartic such as magnesium citrate or sorbitol. Also, treatment may involve careful gastric lavage if performed soon after ingestion or in patients who are comatose or at risk of convulsing. Protect the airway by cuffed endotracheal intubation or by placement of the body in a Trendelenburg and left lateral decubitus position. Obtain chest X-ray and liver function tests. Monitor for cardiac function, respiratory distress and arterial blood gases in severe exposure cases.

**Epinephrine and other sympathomimetic drugs** may initiate cardiac arrhythmias (irregular beating) in persons exposed to high concentrations of this material (e.g., in enclosed spaces or with deliberate abuse). If used, monitor heart action closely. Consider use of other drugs with less arrhythmogenic potential.

# 5. Fire Fighting Measures

NFPA Flammability Classification

Flammable Liquid.

**Flash Point Method** 

CLOSED CUP: 30 ° F. (4 ° C.). (Setaflash)

Flammable Limits

Lower: App. 1.1% Upper: App. 7.1%

Autoignition Temperature

No Data

Hazardous Combustion Products

Burning or excessive heating may produce smoke, carbon monoxide, carbon dioxide, oxides of nitrogen, and possibly other harmful gasses and/or vapors.

Fire and Explosion Hazards

**Flammable Liquid!** This material releases vapors at or below ambient temperatures. When mixed with air in certain proportions and exposed to an ignition source, its vapor can cause a flash fire. Use only with adequate ventilation. Vapors are heavier than air and may travel long distances along the ground to an ignition source and flash back. May create vapor/air explosion hazard in confined spaces such as sewers. If container is not properly cooled, it can rupture in the heat of a fire. Never use welding or cutting torch on or near drum (even empty) because product (even just residue) can ignite explosively.

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**Extinguishing Media** 

Use dry chemicals, carbon dioxide (CO2), foam, water fog, or inert gas (nitrogen). Water fog and spray are effective in cooling containers and adjacent structures by might cause frothing and/or may not achieve extinguishment. A water jet may be used to cool the vessel's external walls to prevent pressure build-up, Autoignition, or explosion. NEVER use a water jet directly on the fire

because it may spread the fire to a larger area.

**Fire Fighting Instructions** 

Product and/or burning liquid will float on water. NEVER use a water jet directly on the fire because it may spread the fire to a larger area. Wear a self-contained breathing apparatus with a full facepiece operated in the positive pressure demand mode with appropriate turn-out gear and chemical resistant personal protective equipment. Refer to the personal protective equipment section of this MSDS. Evacuate area and fight the fire from a maximum distance or use unmanned hose holders or monitor nozzles. Cover pooling liquid with foam. Containers can build pressure if exposed to radiant heat; cool adjacent containers with flooding quantities of water until well after the fire is out. Withdraw immediately from the area if there is a rising sound from venting safety devices or discoloration of vessels, tanks, or pipelines. Notify appropriate authorities if liquid(s) enter sewers / waterways.

#### **Accidental Release Measures**

Take proper precautions to ensure your own health and safety before attempting spill control or clean-up. For more specific information, refer to the Emergency Overview on Page 1, Exposure Controls and Personal Protection in Section 8 and Disposal Considerations in Section 13 of this MSDS.

General

FLAMMABLE LIQUID! Release causes an immediate fire or explosion hazard. Stop the leak if it can be done without risk. Evacuate all non-essential personnel from immediate area and establish a "regulated zone" with site control and security. Do not touch or walk through spilled material. CAUTION: SLIPPERY. Released material will be slippery on smooth, hard surfaces, especially when wet. All equipment used when handling this material should be grounded. Absorb or cover with dry earth, sand, or other non-combustible material, and transfer to appropriate waste containers. Use clean, non-sparking tools to collect absorbed material.

**Small Spills** 

Absorb liquid on vermiculite, floor absorbent, or other absorbent material and transfer to container for disposal.

**Large Spills** 

Secure area and control access. Dike far ahead of a liquid spill to ensure complete collection. Water mist or spray may be used to reduce or disperse vapors; but, it may not prevent ignition in closed spaces. This material will float on water, and its run-off may create an explosion or fire hazard. Eliminate all ignition sources (flares, flames including pilot lights, electrical sparks). Persons not wearing protective equipment should be excluded from area of spill until clean-up has been completed. Stop spill at source, prevent from entering drains, sewers, streams or other bodies of water. Prevent from spreading. If run-off occurs, notify proper authorities that a spill has occurred. Pump or vacuum transfer spilled product to clean containers for recovery. Absorb unrecoverable product. Transfer contaminated absorbent, soil and other materials to containers for disposal.

#### 7. **Handling and Storage**

Handling

A spill or leak can cause an immediate fire / explosion hazard. Keep containers closed and do not handle or store near heat, sparks, or any other potential ignition sources. Bond and ground all equipment before transferring this material from one container to another. Do not contact with oxidizable materials. Do not breathe vapor. Use only with adequate ventilation / personal protection. Never siphon by mouth or take internally. Avoid contact with eyes, skin and clothing. Prevent contact with food, chewing, or smoking materials.

When performing repairs and maintenance on contaminated equipment, keep unnecessary persons away from the area. Eliminate all potential ignition sources. Drain and purge equipment, as necessary, to remove material residues. Use gloves constructed of impervious materials and protective clothing if direct contact is anticipated. Provide ventilation to maintain exposure potential below applicable exposure limits. Promptly remove contaminated clothing. Wash exposed skin thoroughly with soap and water after handling.

Empty containers may contain material residues which can ignite with explosive force. Misuse of empty containers can be dangerous if used to store toxic, flammable, or reactive materials. Cutting or welding of empty containers can cause fire, explosion, or release of toxic fumes from residues. Do not pressurize or expose empty containers to open flame, sparks, or heat. Keep container closed and drum bungs in place. All label warnings and precautions must be observed.

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Return empty drums to a qualified reconditioner. Consult appropriate federal, state and local authorities before reusing, reconditioning, reclaiming, recycling, or disposing of empty containers and/or waste residues of this material. Emergency eyewash fountains and safety showers should be available in the immediate vicinity of potential exposure.

**Storage** 

Store and transport in accordance with all applicable laws. Keep containers tightly closed and store in a cool, dry, well-ventilated place, plainly labeled, and out of closed vehicles. Keep away from all ignition sources! Ground all equipment containing this material. Containers should be able to withstand pressures expected from warming and cooling in storage. This product should be stored in a cool, well-ventilated area. All electrical equipment in areas where this material is stored or handled should be installed in accordance with applicable requirements of the NFPA's National Electrical Code (NEC).

## 8. Exposure Controls and Personal Protection

**Engineering Controls** 

Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapor and/or mists below the pertinent exposure limits (see below). All electrical equipment should comply with the NFPA NEC Standards. Ensure that an emergency eye wash station and safety shower are near the work-station location.

Personal Protective Equipment

Personal protective equipment should be selected based upon the conditions under which this material is used. A hazard assessment of the work area for PPE requirements should be conducted by a qualified professional pursuant to OSHA regulations. The following pictograms represent the minimum requirements for personal protective equipment. For certain operations, additional Personal Protective Equipment may be required.







**Eye Protection** 

Safety glasses with side shields are recommended as a minimum protection. During transfer operations or when there is a likelihood of misting, splashing, or spraying, chemical goggles and face shield should be worn. Suitable eye wash equipment should be readily available.

**Hand Protection** 

Avoid skin contact and use gloves (disposable PVC, neoprene, Nitrile, vinyl, or PVC/NBR). Before eating, drinking, smoking, use of toilet facilities, or leaving work, wash hands with plenty of mild soap and water. DO NOT use gasoline, kerosene, other solvents, or harsh abrasive skin cleaners.

**Body Protection** 

Avoid skin contact. It is recommended that fire-retardant garments (e.g. Nomex <sup>tm</sup>) be worn while working with flammable and combustible liquids. If splashing or spraying is expected, chemical-resistant protective clothing (Tyvek <sup>®</sup>, Nitrile, or neoprene) should be worn. This might include long sleeves, apron, slicker suit, boots, and additional facial protection. If general contact occurs, IMMEDIATELY remove soaked clothing and take a shower. Contaminated leather goods should be removed promptly and discarded.

**Respiratory Protection** 

For unknown vapor concentrations use a positive-pressure, pressure-demand, self-contained breathing apparatus (SCBA). For known vapor concentrations above the occupational exposure guidelines (see below), use a NIOSH-approved organic vapor respirator if adequate protection is provided. Protection factors vary depending upon the type of respirator used. Respirator use should follow OSHA requirements (29 CFR 1910.134) or equivalent standard (e.g. ANSI Z88.2).

**General Comments** 

Warning! Odor is an inadequate warning for hazardous conditions.

# 9. Physical and Chemical Properties

**Physical State** Liquid Color Colorless Aromatic Hydrocarbon Odor pН Not Applicable **Specific Gravity** 0.762 gm./ml. **Liquid Density** 7.36 Lbs. / Gallon **Vapor Pressure** 8 mm Hg @ 68 °F. Vapor Density  $\sim 3.7$  (Air = 1) 251 ° F. < -15 °F. **Boiling Point Freezing Point** Solubility in Water Negligible at 70° F. **Evaporation Rate** 0.76 n-butyl acetate = 1

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## 10. Stability and Reactivity

Chemical Stability Stable

Hazardous Polymerization Not expected to occur.

**Conditions to Avoid** Keep away from extreme heat, sparks, open flame, and strongly oxidizing conditions.

Materials Incompatibility Strong acids, alkalis, and oxidizers such as liquid chlorine, other halogens, hydrogen peroxide,

and oxygen.

**Hazardous Decomposition** 

**Products** 

May form carbon monoxide, carbon dioxide, and other oxides of carbon and nitrogen.

# 11. Toxicological Information

**Carcinogenic Potential:** This product does not contain any intentional components at concentrations above 0.1% which are considered carcinogenic by OSHA, IARC, or NTP.

**Toxicity Data** 

WARNING: THE LC50 VALUES HEREUNDER ARE ESTIMATED ON THE BASIS OF A 4-HOUR EXPOSURE. Acute oral toxicity (LD50): 636 mg/kg [Rat]. Acute dermal toxicity (LD50): 14100 mg/kg [Rabbit]. Acute toxicity of the vapor (LC50): 440 24 hours [Mouse].

Special Remarks on other Toxic Effects on Humans:p. 5 Acute Potential Health Effects: Skin: Causes mild to moderate skin irritation. It can be absorbed to some extent through the skin. Eyes: Cauess mild to moderate eye irritation with a burning sensation. Splash contact with eyes also causes conjunctivitis, blepharospasm, corneal edema, corneal abraisons. This usually resolves in 2 days. Inhalation: Inhalation of vapor may cause respiratory tract irritation causing coughing and wheezing, and nasal discharge. Inhalation of high concentrations may affect behavior and cause central nervous system effects characterized by nausea, headache, dizziness, tremors, restlessness, lightheadedness, exhilaration, memory loss, insomnia, impaired reaction time, drowsiness, ataxia, hallucinations, somnolence, muscle contraction or spasticity, unconsciousness and coma. Inhalation of high concentration of vapor may also affect the cardiovascular system (rapid heart beat, heart palpitations, increased or decreased blood pressure, dysrhythmia, ), respiration (acute pulmonary edema, respiratory depression, apnea, asphyxia), cause vision disturbances and dilated pupils, and cause loss of appetite. Ingestion: Aspiration hazard. Aspiration of Toluene into the lungs may cause chemical pneumonitis. May cause irritation of the digestive tract with nausea, vomiting, pain. May have effects similar to that of acute inhalation. Chronic Potential Health Effects: Inhalation and Ingestion: Prolonged or repeated exposure via inhalation may cause central nervous system and cardiovascular symptoms similar to that of acute inhalation and ingestion as well liver damage/failure, kidney damage/failure (with hematuria, proteinuria, oliguria, renal tubular acidosis), brain damage, weight loss, blood (pigmented or nucleated red blood cells, changes in white blood cell count), bone marrow changes, electrolyte imbalances (Hypokalemia, Hypophostatemia), severe, muscle weakness and Rhabdomyolysis. Skin: Repeated or prolonged skin contact may cause defatting dermatitis.

## 12. Ecological Information

**Ecotoxicity** 

Ecological effects testing has not been conducted on this material. If spilled, this material, its storage tank water bottoms and sludge, and any contaminated soil or water may be hazardous to human, animal, and aquatic life. The petroleum distillates content of this product is volatile and might contribute to the creation of atmospheric smog. Toluene has a half-life of from 3 hours to slightly over 1 day when hydroxyl radicals are present.

**Environmental Fate** 

This material is potentially toxic to freshwater and saltwater ecosystems. It will normally float on water with its lighter components evaporating rapidly. In stagnant or slow-flowing waterways, a hydrocarbon layer can cover a large surface area. As a result, this covering layer might limit or eliminate natural atmospheric oxygen transport into the water. With time, if not removed, oxygen depletion in the waterway might be enough to cause a fish kill or create an anaerobic environment. This coating action can also be harmful or fatal to plankton, algae, aquatic life, and water birds.

For additional ecological information concerning components of this product, users should refer to the Hazardous Substances Data Bank R and the Oil and Hazardous Materials / Technical Assistance Data System (OHM/TADS) maintained by the U.S. National Library of Medicine. (See Section 2 for components.)

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## 13. Disposal Considerations

Waste Management Information

Dispose of in accordance with all applicable local, state, and federal regulations. Recovered non-usable material may be regulated by US EPA as a hazardous waste due to its ignitability (D001) and/or its toxic (D018) characteristics. In addition, conditions of use may cause this material to become a hazardous waste, as defined by Federal or State regulations. It is the responsibility of the user to determine if the material is a RCRA "hazardous waste" at the time of disposal. Transportation, treatment, storage, and disposal of waste material must be conducted in accordance with RCRA regulations (see 40 CFR Parts 260 through 271). State and/or local regulations might be even more restrictive. Contact the RCRA/Superfund Hotline at (800) 424-9436 or your regional US EPA office for guidance concerning case specific disposal issues.

## 14. Transport Information

DOT Information - 49 CFR 172.101

Proper Shipping Name UN 1993, Flammable Liquid, n.o.s., (toluene, heptane, methanol) 3, PG II

ER 128

Hazard Class 3: Flammable Liquid

Packing Group ||

**UN / NA ID** UN 1993

NOS Component toluene, heptane, methanol

RQ (Reportable Quantity) -

49 CFR 172.101

**ERG No.** 128

**Placards Required** 



## 15. Regulatory Information

**TSCA Inventory** The intentional ingredients of this product are listed.

OSHA Process Safety Management – 29 CFR 1910.119 None Listed

**EPA Accidental Release** 

Prevention – 40 CFR 68.130 None Listed

**New Jersey RTK Label** 

Information

See section 2

Pennsylvania RTK Label

Information

See section 2

## 16. Other Information

**Disclaimer of Liability:** 

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#### Abbreviations:

 $\mathsf{App.} = \mathsf{Approximately} \qquad \mathsf{EQ} = \mathsf{Equal} \qquad \mathsf{>} = \mathsf{Greater\ Than} \qquad \mathsf{<} = \mathsf{Less\ Than} \qquad \mathsf{N/AP} = \mathsf{Not\ Applicable} \qquad \mathsf{ND} = \mathsf{No\ Data}$ 

NE = Not Established

ACGIH = American Conference of Governmental Industrial Hygienists

IARC = International Agency for Research on Cancer

NIOSH = National Institute of Occupational Safety and Health

NPCA = National Paint and Coating Manufacturers Association

NFPA = National Fire Protection Association

AIHA = American Industrial Hygiene Association

NTP = National Toxicology Program

OSHA = Occupational Safety and Health Administration

HMIS = Hazardous Materials Information System

EPA = Environmental Protection Agency

#### **Explanation of the HMIS® Ratings**

#### HMIS® III - HEALTH HAZARD RATINGS

- \* Chronic Hazard Chronic (long-term) health effects may result from repeated overexposure
- **0 Minimal Hazard** No significant risk to health
- 1 Slight Hazard Irritation or minor reversible injury possible
- 2 Moderate Hazard Temporary or minor injury may occur
- 3 Serious Hazard Major injury likely unless prompt action is taken and medical treatment is given
- 4 Severe Hazard Life-threatening, major or permanent damage may result from single or repeated overexposures

#### HMIS® III - FLAMMABILITY RATINGS

- **0 Minimal Hazard** Materials that will not burn
- **1 Slight Hazard** Materials that must be preheated before ignition will occur. Includes liquids, solids and semi solids having a flash point above 200 F. (Class IIIB)
- 2 Moderate Hazard Materials which must be moderately heated or exposed to high ambient temperatures before ignition will occur. Includes liquids having a flash point at or above 100 F but below 200 F. (Classes II & IIIA)
- 3 Serious Hazard Materials capable of ignition under almost all normal temperature conditions. Includes flammable liquids with flash points below 73 F and boiling points above 100 F. as well as liquids with flash points between 73 F and 100 F. (Classes IB & IC)
- **4 Severe Hazard** Flammable gases, or very volatile flammable liquids with flash points below 73 F, and boiling points below 100 F. Materials may ignite spontaneously with air. (Class IA)

### HMIS® III - PHYSICAL HAZARD RATINGS

- **0 Minimal Hazard** Materials that are normally stable, even under fire conditions, and will NOT react with water, polymerize, decompose, condense, or self-react. Non-Explosives.
- **1 Slight Hazard** Materials that are normally stable but can become unstable (self-react) at high temperatures and pressures. Materials may react non-violently with water or undergo hazardous polymerization in the absence of inhibitors.
- 2 Moderate Hazard Materials that are unstable and may undergo violent chemical changes at normal temperature and pressure with low risk for explosion. Materials may react violently with water or form peroxides upon exposure to air.
- 3 Serious Hazard Materials that may form explosive mixtures with water and are capable of detonation or explosive reaction in the presence of a strong initiating source. Materials may polymerize, decompose, self-react, or undergo other chemical change at normal temperature and pressure with moderate risk of explosion.
- **4 Severe Hazard** Materials that are readily capable of explosive water reaction, detonation or explosive decomposition, polymerization, or self-reaction at normal temperature and pressure.